

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Application of
Applicant(s) : Faur-Ghenciu et al.
Serial No. : 10/617,146
Filed : July 10, 2003
Title : HIGH ACTIVITY WATER GAS SHIFT CATALYSTS
BASED ON PLATINUM GROUP METALS AND
CERIUM-CONTAINING OXIDES
Docket No. : GMC 0025 PA / 42320.29/GP-302809
Examiner : K. Handal
Art Unit : 1797

Assistant Commissioner for Patents
Washington, D.C. 20231

Sir:

**DECLARATION OF ANCA FAUR-GHENCIU, NATHAN E. TRUSTY,
MARK R. FEAVIOUR, JESSICA G. REINKINGH, PHILLIP SHADY, AND
PAUL J. ANDERSEN UNDER 37 C.F.R. 1.131**

Anca Faur-Ghenciu, Nathan E. Trusty, Mark R. Feaviour, Jessica G. Reinkingh, Phillip Shady, and Paul J. Andersen, the applicants in the above-identified patent application, declare as follows:

1. We are the inventors of claims 1-61 of the above-identified patent application and inventors of the subject matter described and claimed therein.
2. Prior to May 9, 2001, we reduced the present invention to practice as evidenced by Exhibits A-D attached hereto.
3. Exhibit A is a copy of pages 131, 133, 136, 137, 139-141, 161, and 167 of Laboratory Notebook No. 1875. These pages show the preparation of several catalysts of the present invention. Pages 131, 133, 136, 137, 139, 140, 141, and 161 have adhesive labels attached showing receipt of samples for testing by Johnson Matthey Analytical Services. Exhibit

Serial No. 10/617146

Docket No. GMC 0025 PA/40320.29/GP-302809

A has been redacted to delete dates and other proprietary information.

4. Exhibit B is a copy of an email from Anca Ghenciu to Stephen Bransfield and Nathan Trusty, with a cc to Coral Isikci. There is an attachment showing the Powder Catalyst List and Test Procedure. The list shows the Catalyst ID, which corresponds to the page number from the laboratory notebook, the catalyst composition, and the FPR number. The FPR number is assigned when the sample is tested. Exhibit B has been redacted to delete dates and other proprietary information.

5. Exhibit C is a copy of an email from Anca Ghenciu to Stephen Bransfield, with a cc to Sailesh Mullapudi and Nathan Trusty. There is an attachment showing an updated Powder Catalyst List and Test Procedure with the Catalyst ID, the catalyst composition, and the FPR number. Exhibit C has been redacted to delete dates and other proprietary information.

6. Exhibit D is a copy of an email from Anca Ghenciu to Stephen Bransfield and Nathan Trusty. Two attachments show the Powder Catalyst List and Test Procedure with the Catalyst ID, the catalyst composition, and the FPR number. There are also attachments of graphs showing the test results for various samples. Exhibit D has been redacted to delete dates and other proprietary information.

7. Exhibit E shows the correlation between the Catalyst ID, the FPR number, and the catalyst compositions shown in Exhibits A-D.

8. Each of the dates deleted from Exhibits A-D is prior to May 9, 2001. All work relating to the conception and reduction to practice of this invention was carried out in a WTO country.

Serial No. 10/617146

Docket No. GMC 0025 PA/40320.29/GP-302809

The declarants further state that the above statements were made with the knowledge that willful false statements and the like are punishable by fine and/or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of this application or any patent resulting therefrom.

Date: March 4, 2008.

Anca Faur-Ghenciu

Date: _____

Nathan E. Trusty

Date: _____

Mark. R. Feaviour

Date: _____

Jessica G. Reinkingh

Date: _____

Phillip Shady

Date: _____

Paul J. Andersen

Serial No. 10/617146

Docket No. GMC 0025 PA/40320.29/GP-302809

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Date: _____

Date: 3-4-08

Date: _____

Date: _____

Date: _____

Date: _____

Anca Faur-Ghenciu

Nathan E. Trusty
Nathan E. Trusty

Mark. R. Feaviour

Jessica G. Reinkingh

Phillip Shady

Paul J. Andersen

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Date: _____

Anca Faur-Ghenciu

Date: _____

Nathan E. Trusty

Date: 3/3/08

M. Feavour

Mark. R. Feaviour

Date: _____

Jessica G. Reinkingh

Date: _____

Phillip Shady

Date: _____

Paul J. Andersen

Serial No. 10/617146

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Date: _____

Anca Faur-Ghenciu

Date: _____

Nathan E. Trusty

Date: _____

Mark R. FeaviourDate: March 8, 2008

Jessica G. Reinkingh

Date: _____

Phillip Shady

Date: _____

Paul J. Andersen

Serial No. 10/617146

Docket No. GMC 0025 PA/40320.29/GP-302809

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Date: _____

Anca Faur-Ghenciu

Date: _____

Nathan E. Trusty

Date: _____

Mark. R. Feaviour

Date: _____

Jessica G. ReinkinghDate: March 4, 2008

Phillip Shady

Date: _____

Paul J. Andersen

Serial No. 10/617146

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Date: _____

Anca Faur-Ghenciu

Date: _____

Nathan E. Trusty

Date: _____

Mark. R. Feaviour

Date: _____

Jessica G. Reinkingh

Date: _____

Phillip Shady

Date: 3/4/2008

Paul J. Andersen

ITLE La2O3 CeO2 Impregnation

PROJECT NO. In House
BOOK NO. Material

131

Work continued from Page

$$\frac{X}{40} = \frac{1.4839}{1.308}$$

45.6101 Vol

$$I.W.C. = 13008g \text{ Powder}$$

$$DI = 1.4839$$

1757-131-1

$$(40g La_2O_3 CeO_2 + X) \cdot 0.01 = X$$

$$0.4g = X(99)$$

$$X = 404g \text{ Pd} / \frac{14933g \text{ Pt} / 5 \text{ Pt(NO}_3)_2}{150.79} = 2.679$$

$$2.72g \text{ Pd(NO}_3)_2$$

- ① Mix DI + Pd as Nitrat lot # ~~718~~ 718-66666001 to 40.0g Powder as La2O3 CeO3 In Excess. 210%
- ② Mix well and Place In drying oven - for 2hr than fired @ 500°C for 2hr - @ 125°C → 500 10°C/min

Submitter: TRUSTY

Logged:

57-131-1 INHOUSE

Take 40g Powder and @ add to It Pd @ 10% and Pt @ 5% via Co imp.

$$X_{Pd} = 0.404g \text{ Pd} / 0.15079g/g = 2.679g \text{ Pd(NO}_3)_2$$

$$X_{Pt} = 0.202g \text{ Pt} / 0.14933g/g = 1.353g \text{ Pt as Nitrat}$$

$$= 1.8896m^2$$

$$= 0.8655m^2$$

- ① Mix Pt & Pd into DI H2O ~ 42g of DI mix well
- ② Add Soln to Powder. slowly w/ good mixing. Be careful to make sure all powder is blended in w/ PGM. @ 125°C 2hr
- ③ Place In drying oven and fire for 500°C 2hr 10°C/min @ 125 → 500

1757-131-2

Analytical Services

Submitter: TRUSTY

Logged:

VTIFIC BINDERY PRODUCTIONS CHICAGO

ID: INHOUSE 1757-131-2

Sub Id:

continued to Page

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DATE

USED TO AND UNDERSTOOD BY

DATE

WITNESS

DATE

EXHIBIT
A

In house lab

Work continued from Page

190 Pt on .5% Cs

Take 4g of La_2CoO_2 Powder
 + 2.70g of Pt as $\text{Pt}(\text{NO}_3)_2 = 1.8 \text{ ml} - \text{Total} = 4.38 \text{ ml}$
 + $X_{\text{Cs}} = .20 \text{ g Cs}$

$\frac{132.90 \text{ g Cs}}{194.91}$ will use Cs Nitrate instead of Cs Acetate.

$X_{\text{Cs}} = .20 \text{ g Cs} \frac{178.91 \text{ g}(\text{Cs}(\text{NO}_3)_2)}{132.90 \text{ g Cs}} = .293 \text{ g}(\text{Cs}(\text{NO}_3)_2)$
 Act = .381g

+ Mix w/ Total Volume of $\text{Pt}(\text{NO}_3)_2 + \text{DI}$ for a Co Impregnation -

* Note Soln had to be mixed w/ stir plate to dissolve the Cs Nitrate.

+ Mix soln gradually into powder until total mixture is saturated.

+ Place in drying oven @ 125°C for 2 hrs.

Note: During the Dry step take saturated powder out and mix w/ spatula so that wet cake can dry evenly.

Temp step from 0°C @ $26^\circ/\text{min}$. Know oven available downstairs for firing.

23

Lab ID 1757-133-4

Analytical Services

Submitter: TRUSTY

Logged:

ID: INHOUSE 1757-133-4

Sub Id:

SCIENTIFIC BINDERY PRODUCTIONS CHICAGO 60605 MADE IN USA

SIGNATURE

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DATE

WITNESS

Work continued to Page

DATE

DATE

Preps. 1757-136-1

ID: 1757-136-1

Submitter: TRUSTY

Analytical Services

Loaded:

Sub Id:

Iw. Condition

6.2621g 101

2.7818g DI

17.764ml; 40g

Take Pd @ 1% on 40g of

$$(40g 101 (6.2621g) + X) \cdot 0.01 = X$$

$$\frac{0.4g}{0.99} = X_{Pd}$$

17.76 - 1.89

$$X_{Pd} = 404g / 15079 = 2.68g Pd \text{ as nitrate}$$

① Mix (Pd nitrate) w/ DI (17.69g DI Excess)
to 40g

$$Tot wt = 40.15g$$

② Mix well to avoid dry spots.

③ Placed wet cake in dry oven overnight
4:15pm → Till 5:00Am.

④ Remove and place in firing oven @ 500°C 2hr.

Note: It's important to Break up cake prior to firing.

M. Tandy

From: Anca Ghenciu
To: Trusty, Nathan
Date:
Subject: Impregnation

Nathan,

30-50 g of each, depending on how much powder we have.
I would make them by co-impregnation, but successive impreg is ok

- 1 1%Pd/La2O3-CeO2
- 2 1%Pd-0.5%Pt/La2O3-CeO2
- 3 1%Pt-0.5%Cu/La2O3-CeO2
- 4 1%Pt-0.5%Cs/La2O3-CeO2

Allows powder
to fire evenly.

M. Tandy

Work continued from Page

Catalyst ID 1757-137-2 [190 Pd - 590 Pt/]

Pd:Pt: Done as Co-~~P~~ Impregnation - 39.50g Tot. After 500°C

① Take 40g of ... and P/L to in to container

② w/ I.W. being 17.76ml total Volume in Excess
see I.W. condition on 1757-136
$$X_{Pd} = .404g Pd / .15079g/g = 2.68g Pd as Nitrate$$

$$X_{Pt} = .202g Pt / .14933g/g = 1.352g Pt as Nitrate$$

 15 Pt as $Pt(NO_3)_2$ Tot Volume of PGM = 2.75ml - 17.76ml Tot vol
 lot-71866666001 of I.W. cond.

Tot DI Needed = 15.00ml

③ Add DI of 15ml to PGM solution - [mix well]

④ Take ... and Gradually mix soln slowly

⑤ Add all soln to ... [until compl] make sure
there are no clumps in saturated mixture.

⑥ Place in drying oven @ 125°C 8:15am

 25 ⑦ In Bf oven @ 10:15am. 2hr @ 500°C ← Brk cake up
 prior to firing using a spatula.

VS In House

 Dissolved
 better in
 solutions
 darker in color

Analytical Services

Submitter: TRUSTY

Logged:

ID: 1757-137-2

Sub Id:

SIGNATURE

DATE

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WITNESS

DATE

TITLE Pt on In house

PROJECT NO.

139

LaCe Material @ 17%

BOOK NO.

Work continued from Page

Pt on LaCe

1757-139-2

Total vol = 17.76ml

① Take 40g of In House Material

$$4g = X(99)$$

$$S = 1.5632$$

$$X_{Pt} = 40g / 1.4933g/gPt(NO_3)_2 = 2.679g Pt(NO_3)_2$$

② Take 17.76ml DI - Vol of Pt = 16.05ml DI
Mix DI + Pt together.③ Add Pt + DI to In house LaCe Oxide gradually.
* Wet cake will form. Very hard to mix w/spat.
* Continue Adding mixture to powder until
saturate completely and all soln. is in.④ Be careful that all powder is mixed ~~tho~~ well.
(powder) cake has tendency to avoid soln by hiding in
pocket.⑤ Place wet cake in Drying oven @ 125°C
This cake was left over night prior to firing
(will be)⑥ Place Dried cake in to oven. prior to firing
dried cake break up wet cake well.
600°C for 2hr.

Analytical Services

Submitter: TRUSTY

Logged:

ID: INHOUSE 1757-139-2

Sub Id:

SCIENTIFIC BINDERY PRODUCTIONS CHICAGO 60605 MADE IN USA

Work continued to Page

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DATE

WITNESS

DATE

DATE

Impregnation Using Pt @ 1%

Pt @ Nitrate C

① 40g of material and place into crucible

② $X_{Pt} = 40g / .14933 = 2.705g \text{ Pt}$ or 1.23ml
384g

Same Lot

7186666601

Lot concentration

149.33g/kg

S = 1.5631

Mix Pt Nitrate + DI H₂O to a total volume of 17.76ml to be ~10% Excess soln.③ Add 16.029ml of DI H₂O to Pt(NO₃)₂④ Begin adding soln to powder gradually will become hard but once it is saturated make sure all clumps are crushed to avoid any powder from be catalyzed evenly. ~~Have~~⑤ Place in drying oven of 125°C for 2 hr
Turner of 10 min

Cat ID #1757-140-1

⑥ Place in oven @ 500°C for 2 hr

Analytical Services

Submitter: TRUSTY

Logged:

ID:

PT:

1757-140-1

Sub Id:

m
S. J. J. J.

TITLE Impregnation of

PROJECT NO.

141

PTQ 190 CS @ .590 on/b

BOOK NO.

(RP)

Work continued from Page

Take 40g ~~24~~

Lab from (RP)

lot ~~88~~ 98160/98

$$X_{Pt} = 404 / .14933 = 2.705g Pt(NO_3)_2$$

$$X_{CS} = .202CS / \frac{194.91CS(NO_3)_2}{132.91CS} = \boxed{2.962g CS(NO_3)_2}$$

① Take $CS(NO_3)_2$ + (16.029 g DI H_2O and ~~DI~~) mix by stir plate until dissolved completely.

~ 3 minutes to dissolve. Soln crashed
Note order of Addition ~~if~~ $DDI + CS$
~~DI~~ (A + DI) + $CS(NO_3)_2$ are placed together and ~~DI~~ Pt added to.

② Once Mixture is thoroughly mixed add

to
③ mixture ~~is~~ will visually appear to be liquid.
Remember this is in excess soln.

④ Drying oven @ 125°C 2 hr Time = 10:00 Am.

1757-141-1 I.D.#

Analytical Services

Submitter: TRUSTY

Logged:

ID:

PT:CS 1757-141-1

Sub Id:

SCIENTIFIC BINDERY PRODUCTIONS CHICAGO 60605 MADE IN USA

Work continued to Page

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WITNESS

DATE

20% Pt / Cs Nitrate 20%

Work continued from Page

① Take 1757-159 and Grind material into fine powder

② Take 40g of 1757-159 (20% Pt only)
see Cs calculation on Pg 160B.Take 118g Cs Nitrate and Dissolve it in 17.26g DI
③ Dissolve $\text{Cs}(\text{NO}_3)_2$ + DI and Add to Pt.
powder slowly.④ Once soln. is completely added to powder as will
Good Homogeneity Place in drying oven 2hr.

⑤ Fire @ 500°C 2hr.

Cal ID = 1757-161A

Take and Place 20% Pt By IW.

IW of

$$\frac{2.2479}{1.4877} = 98$$

$$X = 62.87$$

Take 95g and Place 20% $\text{Pt}(\text{NO}_3)_2$ to soln. to it.

$$95g + X_{\text{Pt}} = 1.9987g / 1.4399 = 13.46g \text{ Pt}(\text{NO}_3)_2$$

$$= 8.61\text{ml} - 62.87 = 54.2\text{ml DI.}$$

① Mix DI H_2O + PGM well② Add PGM + DI to Gradually until all soln
is mixed in.

a) overshoot/calculation error for IW condition.

Placed in Air Drying oven Down starts @ 125°C

Dried well w/ intermittent stirring.

③ Place in firing oven @ 125-500°C Cal ID =
2hr. 1757-161B

N. T. Smith

J. P. Smith

Shady Powders

Work continued from Page

Inv of

$$\frac{2.5284g}{1.3802g \text{ DE}}$$

$$\frac{30g}{x}$$

$$TV = 15.90g \text{ DE}$$

or 14.60 + 1.6m.

Take 30g of material. and Place 10/6/6 & Sample or

$$N-Pt = .303g / .14933 = 2.029g \text{ Pt}(\text{NO}_3)_2 \text{ or } 1.298 \text{ ml}$$

① Place $\text{Pt}(\text{NO}_3)_2 + \text{DI H}_2\text{O}$ on ~~to~~ 30g of slowly. Making sure all ~~of~~ ^{is} soln is evenly placed on powder.

② After adding all soln. Place wet cake in/d drying oven @ 125°C. This powder will be in oven up to 15. overnight.

③ Next Day. Placed in oven for 2 hr @ 500°C
Starting Time 5:23 Am.
Assay # 35147

1757-167A

Cs on 206 Pt.

Take 1757-166A and Place Cs EX

④ Take 41.4g Amol. Pt2

$$TV = 27.39 \text{ ml} \quad C = .002$$

$$X_{Cs} = .08296 / \frac{194.91 \text{ Cs}(\text{NO}_3)_2}{132.91 \text{ Cs}} = .12167g \text{ Cs}(\text{NO}_3)_2$$

Act = .1204g

① See Procedure 1757-166A.

Note: This material will be # Cs/Pt.

a) Impreg's w/ Buffer Cs @ 150g/l

c) Pt (Ex) Dry fine

b) Cs EX Dry fine.

Assay #

1757-167B

SCIENTIFIC BINDER PRODUCTIONS CHICAGO 60605 MADE IN USA

Work continued to Page

SIGNATURE

DATE

DISCLOSED TO AND UNDERSTOOD BY

DATE

WITNESS

DATE

J. Green

Prior, Patricia

From: Anca Ghenciu [ahencia.EP.DEVON@matthey.com]
Sent:
To: Stephen Bransfield; Nathan Trusty
Cc: Coral Isikci
Subject: Powder tests

Attachments: RxnTest_SCAT-FC_F .doc



RxnTest_SCAT-FC_

Steve, Nathan,

Although it seems odd, I am writing all these messages now (Saturday night) because on Monday morning by the time I get here you will be having many things already achieved and I may be too late.

For the powder reactor, there are four tests I would like to have run before I leave to UK (see Table attached): lines 13, 21, 20, 23, run in this order. You may have already tested line #13, I am not sure since I do not have the data for that. Could you test these on Monday and Tuesday AM? If not, whatever it is possible.

There is no need to repeat 1%Pt/CeO₂ (C480-51B, Sonning) (it used to be line #16) for the time being, but if it is possible, after you set-up the GC, I would like to have the following tested using both MS and GC, so that we can validate the GC for the powder reactor:

- Cu-Zn powder
- line #8 (cat 1757-137-2); we had postponed this initially
- line#22 (cat C480-75A)
- repeat line #16 (best to date, 1757-142-12)

All above 45-60 mesh, 1 g cat, 1 g cord, 10 slmp total flow, concentrations as before. When you send me the test results, please also copy Nathan form now on, so that we all stay in the loop.

Nathan,

For the above tests, Steve will only need 2 g (1cat+1cord) of 1757-142-12, and also the memorable jar of Cu-Zn and a jar of cordierite, both at 45-60 mesh, so that he can validate the GC with it from time to time.

Thank you very much,

Anca

Powder Catalyst List and Test Procedure, Matrix 3_LT

	Catalyst ID	Catalyst Composition	
		Cu-ZnO commercial (Sud-Chemie)	3FPR95
1	1757-129	MgO-Al ₂ O ₃ (P)/Ni/Fe	3FPR94
2	1757-131-1	1%Pd/75%La ₂ O ₃ -25%CeO ₂	3FPR96
3	1757-131A2	1%Pd-0.5%Pt/75%La ₂ O ₃ -25%CeO ₂	3FPR97
4	1757-132-3	1%Pt-0.5%Cu/75%La ₂ O ₃ -25%CeO ₂	3FPR98
5	1757-133-4	1%Pt-0.5%Cs/75%La ₂ O ₃ -25%CeO ₂	3FPR99
6	1757-139-2	1%Pt/75%La ₂ O ₃ -25%CeO ₂	3FPR104
7	1757-136-1	1%Pd/25%La ₂ O ₃ -75%CeO ₂	3FPR105
8	1757-137-2	1%Pd-0.5%Pt/ 25%La ₂ O ₃ -75%CeO ₂	
9	1757-138-3	1%Pt-0.5%Cu/ 25%La ₂ O ₃ -75%CeO ₂	3FPR106
10	1757-141-1	1%Pt-0.5%Cs/ 25%La ₂ O ₃ -75%CeO ₂	3FPR107
11	1757-140-1	1%Pt/ 25%La ₂ O ₃ -75%CeO ₂	3FPR108
12	1757-142-12	1%Pt/CeO ₂ (3FPR109
13	1757-14x-13	1%Pt-0.5%Cu/CeO ₂	Already tested??
14	1757-143-14	1%Pt-0.5%Cs/CeO ₂	3FPR110
15	1757-144-15	1%Pt-0.5%Cs-0.5%Cu/CeO₂	3FPR111
16	1757-142-12	1%Pt/CeO ₂ (RERUN)	Re-run
17	C480-82	10%(20%Pd/Fe ₂ O ₃)/Al ₂ O ₃	3FPR100
18	C480-74	5%Pd/Fe ₂ O ₃	3FPR102
19	C480-90A	1%Au/TiO ₂ (gray)	3FPR103
20	C480-83	1%Au/TiO ₂ (lilac)	
21	C480-90B	1%Au/1%Co/TiO ₂ (gray)	
22	C480-75A	1%Au/Fe ₂ O ₃	
23	C480-75B	5%Au/Fe ₂ O ₃	
24	C480-71	MoS ₂ /La ₂ O ₃ -Al ₂ O ₃	3FPR70

Test procedure:

1 g catalyst, 1 g cordierite, each sieved to 45-60 mesh, well mixed.

Total flow 1.125 SLPM

Inlet mole %: 8%CO, 30%H₂O, 10%CO₂, 32.5%H₂, 1%CH₄, 18.5% N₂ (balance)

Steady state temperatures: **150**, 175, 200, 225, 250, 275, 300, 350, 400, 450, 500, 550°C.

Please save the spent samples in new vials, with the run # on the vial.

Prior, Patricia

From: Anca Ghenciu [ghencia.EP.DEVON@matthey.com]
Sent:
To: Stephen Bransfield
Cc: Sallesh Mullapudi; Nathan Trusty
Subject: updated table

Attachments: RxnTest_SCAT-FC_ .doc



RxnTest_SCAT-FC_

Steve,

I have updated the Table. Nathan will bring more catalysts tomorrow.

Thanks,

Anca

Powder Catalyst List and Test Procedure, Matrix 4_LT

	Catalyst ID	Catalyst Composition	Test Conditions	Test #
	C18-7		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR122
	C480-96A	1%Pt/La-CeOx	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	Repeat of FPR121 FPR123
	C480-74		(45-60 mesh, 2g catalyst), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150-500°C	Repeat of FPR120 FPR124
	C480-100A		(45-60 mesh, 2 g catalyst), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2.5slpm, 150°-500°C	
	C480-100B		(45-60 mesh, 2 g catalyst), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2.5slpm, 150°-500°C	
	1757-4-149-1A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR126
	1757-4-149-2A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR125
	1757-4-149-3A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR127
	C480-110B		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2 slpm, 150°-500°C	FPR128
	C480-68A		durability test, (45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150-275°C, 3 ramps	FPR129
	C480-110A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2 slpm, 150°-500°C	FPR130
	1875-01		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR131
	1875-02		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR132
	1875-03		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ ,	FPR133

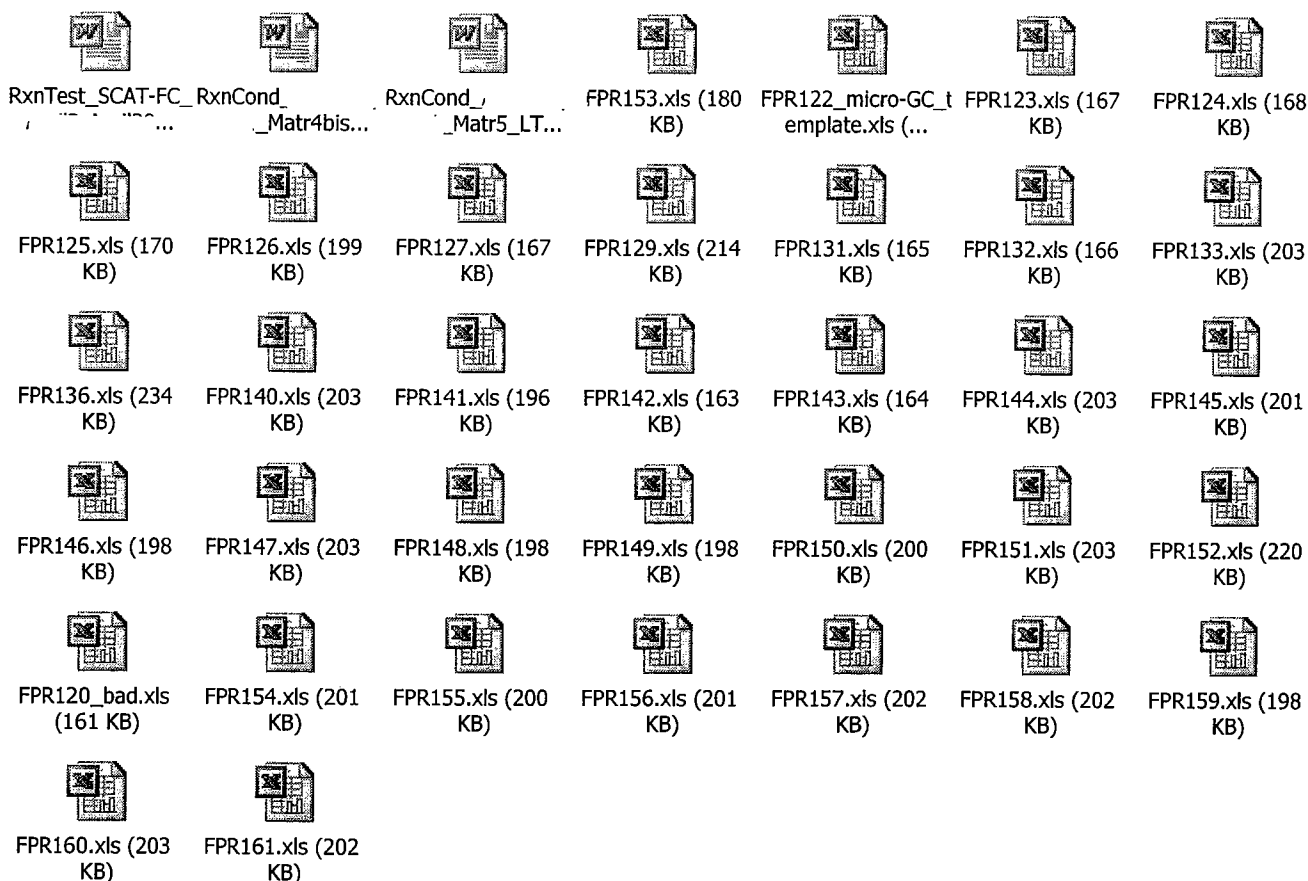
			1.125 slpm, 150°-500°C	
	1875-04		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR136
	C480-112A	Sonning reformer (Li-0.5%Rh/)	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2 slpm, 150°-500°C	FPR135
	C480-112B	Sonning reformer (Cs,0.5%Rh/)	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2 slpm, 150°-500°C	FPR139
	C480-112C	Sonning reformer (Cs,1%Rh/)	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2 slpm, 150°-500°C	FPR138
	1757-159		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR140
	1757-161A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR142
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
Calc 400°C, 2 hr.	1875-09		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
	1757-159B	2%Pt/1%La-CeO₂	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR141
	1757-160B	2%Pt/0.2%Cs/1%La-CeO₂ This is a 0.2%Cs/1757-159B	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR143
Calc 400°C, 2 hr.	1875-10		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
	1757-162A	0.5%Rh/1%La-CeO₂		FPR144
	1757-160A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
	1757-161B		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
	1757-162B		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	

	1875-05		To test later	
	Nathan, please fill in ID		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
Calc. 400°C, 2hr.	1875-07		To test later	
	Nathan		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
	Nathan		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
	Anca		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	

Prior, Patricia

From: Anca Ghenciu [ghencia.EP.DEVON@matthey.com]
Sent:
To: Stephen Bransfield; Nathan Trusty
Subject: Updated files from

Attachments: RxnTest_SCAT-FC_.doc; RxnCond_ _Matr4bis.doc;
RxnCond_ _Matr5_LT.doc; FPR153.xls; FPR122_micro-GC_template.xls;
FPR123.xls; FPR124.xls; FPR125.xls; FPR126.xls; FPR127.xls; FPR129.xls; FPR131.xls;
FPR132.xls; FPR133.xls; FPR136.xls; FPR140.xls; FPR141.xls; FPR142.xls; FPR143.xls;
FPR144.xls; FPR145.xls; FPR146.xls; FPR147.xls; FPR148.xls; FPR149.xls; FPR150.xls;
FPR151.xls; FPR152.xls; FPR120_bad.xls; FPR154.xls; FPR155.xls; FPR156.xls;
FPR157.xls; FPR158.xls; FPR159.xls; FPR160.xls; FPR161.xls



Updated files form

Matrix 4_LT ()
Matrix 5_LT () - current Matrix 4bis_LT (Pt/Cu-Zn UCI)

The attached have the correction for methanation (taking into account the CH₄ forms from CO). Please replace the old versions with these. You can use any of these (for instance, the template) for future tests.

If there is any missing file between FPR120 and FPR161, please let me know.

Anca

Powder Catalyst List and Test Procedure, Matrix 4 LT

Catalyst ID	Catalyst Composition	Test Conditions	Test #
C18-7		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR122
C480-96A	1%Pt/La-CeOx	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	Repeat of FPR121 FPR123
C480-74		(45-60 mesh, 2g catalyst), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	Repeat of FPR120 FPR124
C480-100A		(45-60 mesh, 2 g catalyst), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2.5slpm, 150°-500°C	
C480-100B		(45-60 mesh, 2 g catalyst), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2.5slpm, 150°-500°C	
1757-4-149-1A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR126
1757-4-149-2A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR125
1757-4-149-3A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR127
C480-110B		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2 slpm, 150°-500°C	FPR128
C480-68A		durability test, (45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150-275°C, 3 ramps	FPR129
C480-110A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2 slpm, 150°-500°C	FPR130
1875-01		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR131
1875-02		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR132
1875-03		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ ,	FPR133

			1.125 slpm, 150°-500°C	
	1875-04		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR136
	C480-112A	Sonning reformer (Li-0.5%Rh/)	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2 slpm, 150°-500°C	FPR135
	C480-112B	Sonning reformer (Cs,0.5%Rh/)	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2 slpm, 150°-500°C	FPR139
	C480-112C	Sonning reformer (Cs,1%Rh/)	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 2 slpm, 150°-500°C	FPR138
	1757-159		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR140
	1757-161A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR142
	1757-164B		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
Calc 400°C, 2 hr.	1875-09		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR149
	1757-159B	2%Pt/La-CeO₂	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR141
	1757-160B	0.2%Cs/2%Pt/La-CeO₂ This is a 0.2%Cs/1757-159B	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR143
Calc 400°C, 2 hr.	1875-10		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	
	1757-162A	0.5%Rh/La-CeO₂		FPR144
	1757-160A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR145
	1757-161B	2%Pt	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR146
	1757-163B	2%Pt/0.2%Cs	(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , 1.125 slpm, 150°-500°C	FPR148

	1875-05		To test later	
	1757-162B		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, 1.125 slpm, 150°-500°C	FPR147
Calc. 400°C, 2hr.	1875-07		To test later	
	1757-163A		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, 1.125 slpm, 150°-500°C	
	Nathan		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, 1.125 slpm, 150°-500°C	
	Anca		(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1g +1g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, 1.125 slpm, 150°-500°C	

Powder Catalyst List and Test Procedure, Matrix 5_LT

Rxn FPR152 (1)

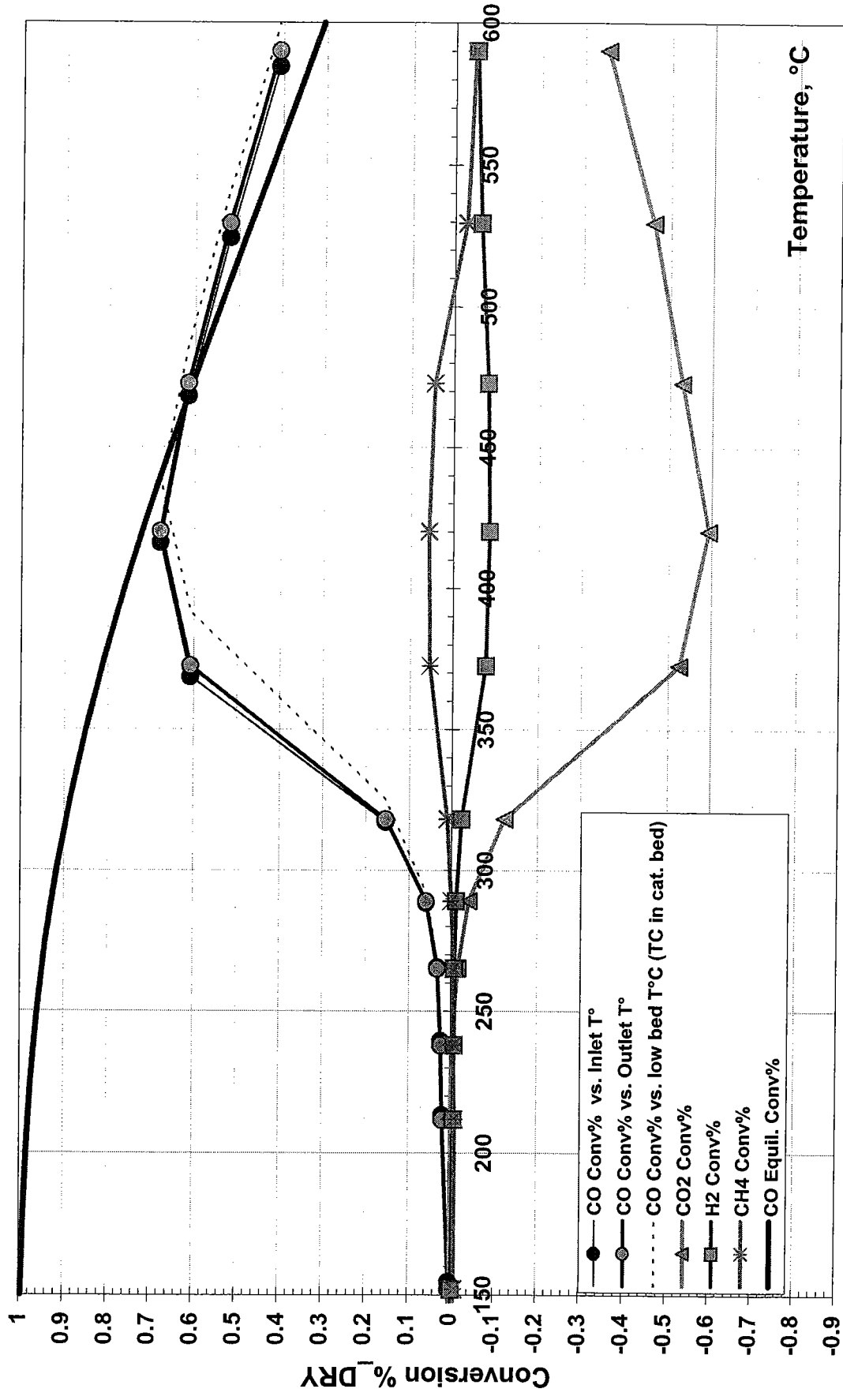
#	Catalyst ID	Catalyst Composition	Test Conditions	Test ID
	C480-100A		(45-60 mesh, 2 g catalyst), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ , 2.5 slpm, 150°-500°C	
	C480-100B		(45-60 mesh, 2 g catalyst), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ , 2.5 slpm, 150°-500°C	
			(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ , 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ , 1.125 slpm, 150°-500°C	
	1875-07		(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ , 1.125 slpm, 150°-500°C	
	1757-163A		(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ , 1.125 slpm, 150°-500°C	FPR151 (already reported to GM at April meeting)
	1757-164A		(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ , 1.125 slpm, 150°-500°C	FPR153
	Prep		(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ , 1.125 slpm, 150°-500°C	
	1757-164B		(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ , 1.125 slpm, 150°-500°C	FPR160
	Improvement			
	1757-166B	(2%Pt-0.2%Cs)/CeO ₂ -ZrO ₂ () (co-impreg, DI???) To be compared with seq. 1757-163B	(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ , 1.125 slpm, 150°-500°C	FPR157
	Prep		(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H ₂ , 30%H ₂ O, 10%CO ₂ , 1%CH ₄ , bal. N ₂ ,	

			1.125 slpm, 150°-500°C	
	Improvement			
1757-166A	2%Pt/CeO2-ZrO2 (from citric acid To be compared with 1757-161B		(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, bal. N2, 1.125 slpm, 150°-500°C	FPR156
1757-167B	0.2%Cs/2%Pt/CeO2-ZrO2 (citric acid (both Pt and Cs)		(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, bal. N2, 1.125 slpm, 150°-500°C	FPR159
Prep			(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, bal. N2, 1.125 slpm, 150°-500°C	
Prep			(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, bal. N2, 1.125 slpm, 150°-500°C	
Prep			(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, bal. N2, 1.125 slpm, 150°-500°C	
	Improvement			
			(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, bal. N2, 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, bal. N2, 1.125 slpm, 150°-500°C	
			(45-60 mesh, 1 g catalyst + 1 g cordierite), 8%CO, 32.5%H2, 30%H2O, 10%CO2, 1%CH4, bal. N2, 1.125 slpm, 150°-500°C	
	Improvement		Phil Shady	
1757-167A				FPR 158
1757-168A				FPR161

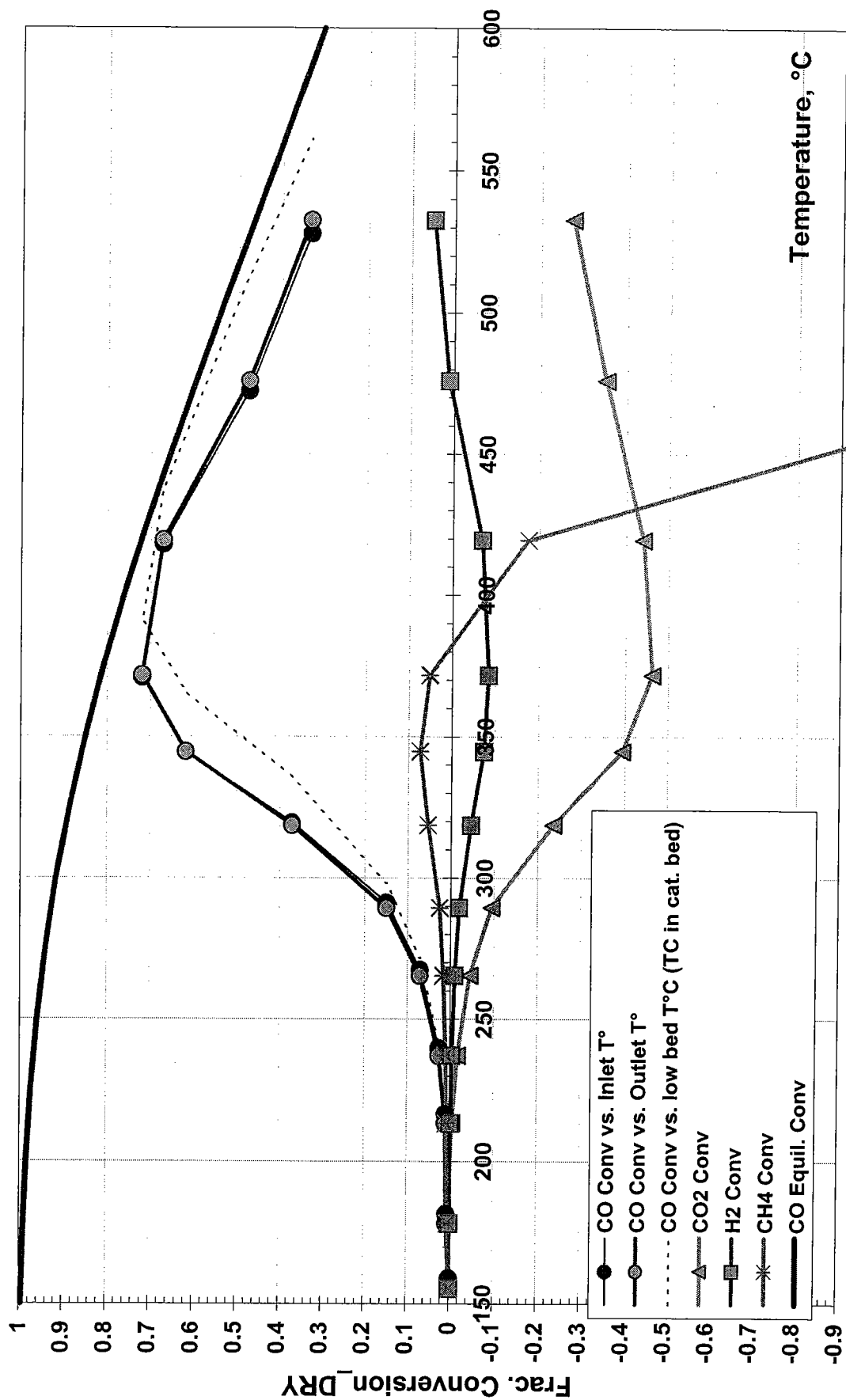
[illegible]

	Materials to be subjected to durability study (3 T ramps, new T schedule, very high T~600°C) - no CH4 in the feed - high SV	- - - - -		

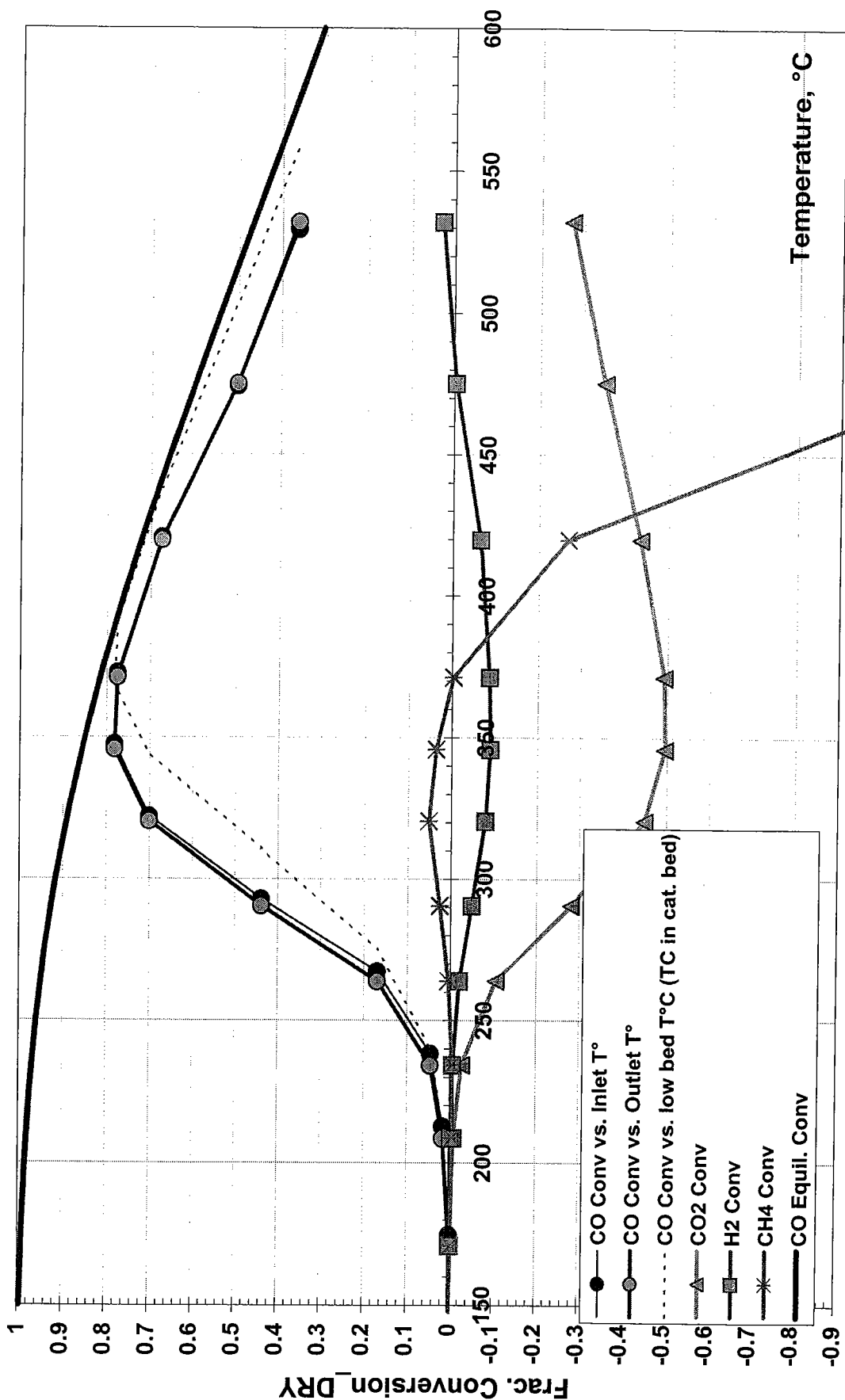
**FPR123 1%Pt/La-CeOx (C480-96A) 1g cordierite/1g cat, 45-60 mesh, Total Flow
1.125slpm**



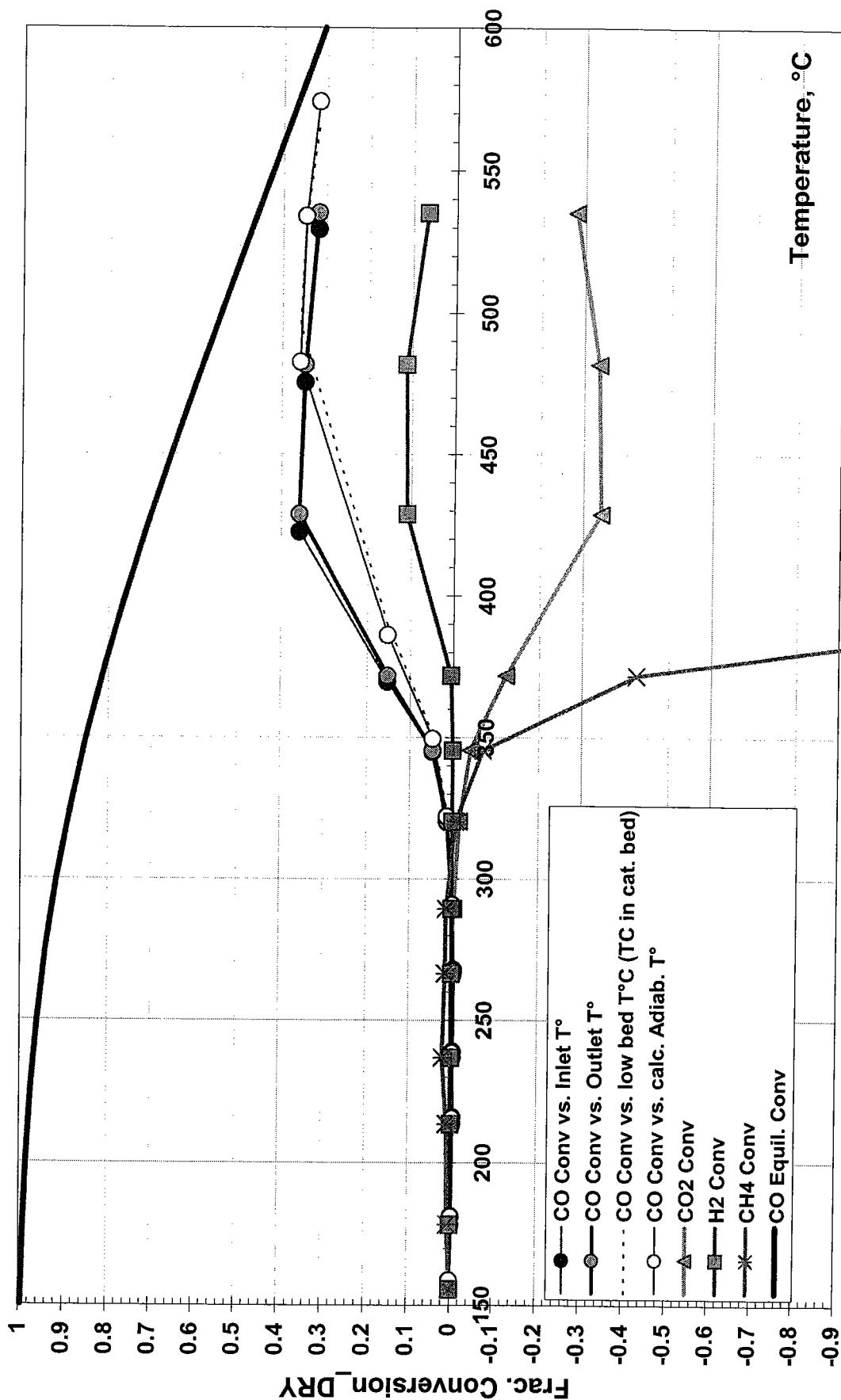
FPR141 2%Pt/La-CeO2 ((1757-159B) 1.0gcat/1.0g cord, 45-60 mesh, Total Flow
1.125slpm

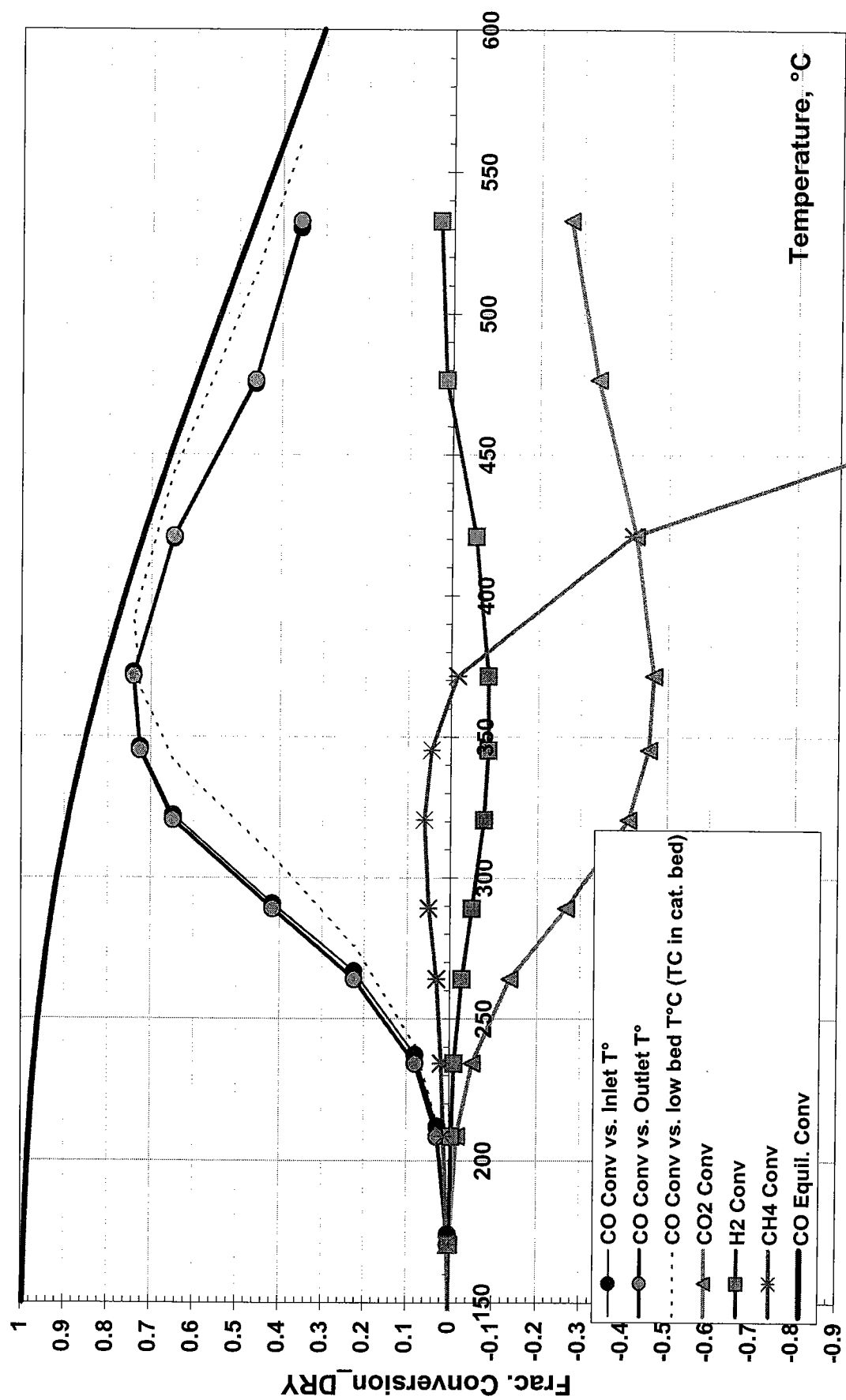


FPR143 0.2%Cs/2%Pt/La-CeO₂ (---) (1757-160B) 1.0gcat/1.0g cord, 45-60 mesh,
Total Flow 1.125slpm

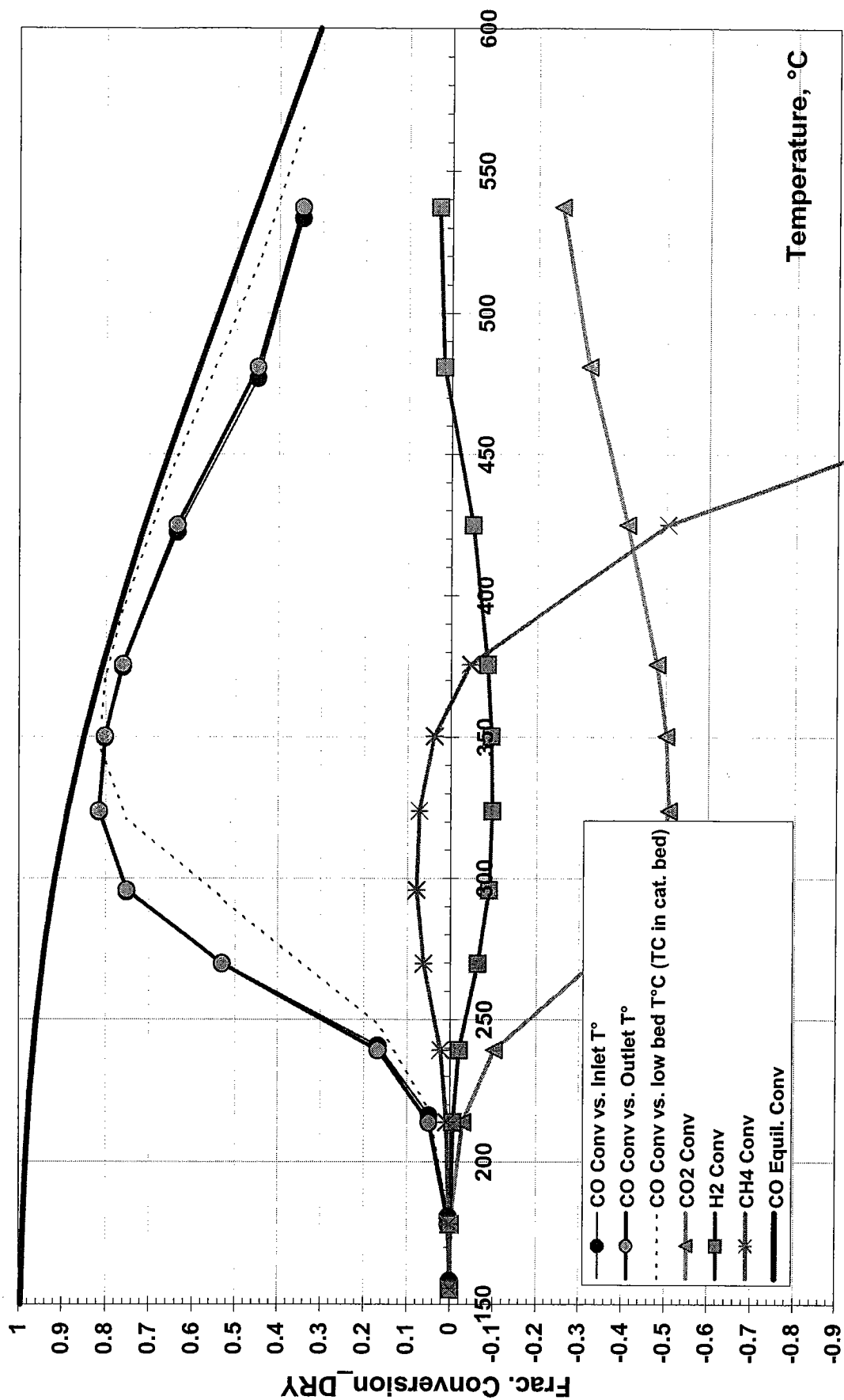


FPR144 0.5%Rh/La-CeO2 () (1757-162A) 1.0gcat/1.0g cord, 45-60 mesh, Total
Flow 1.125slpm

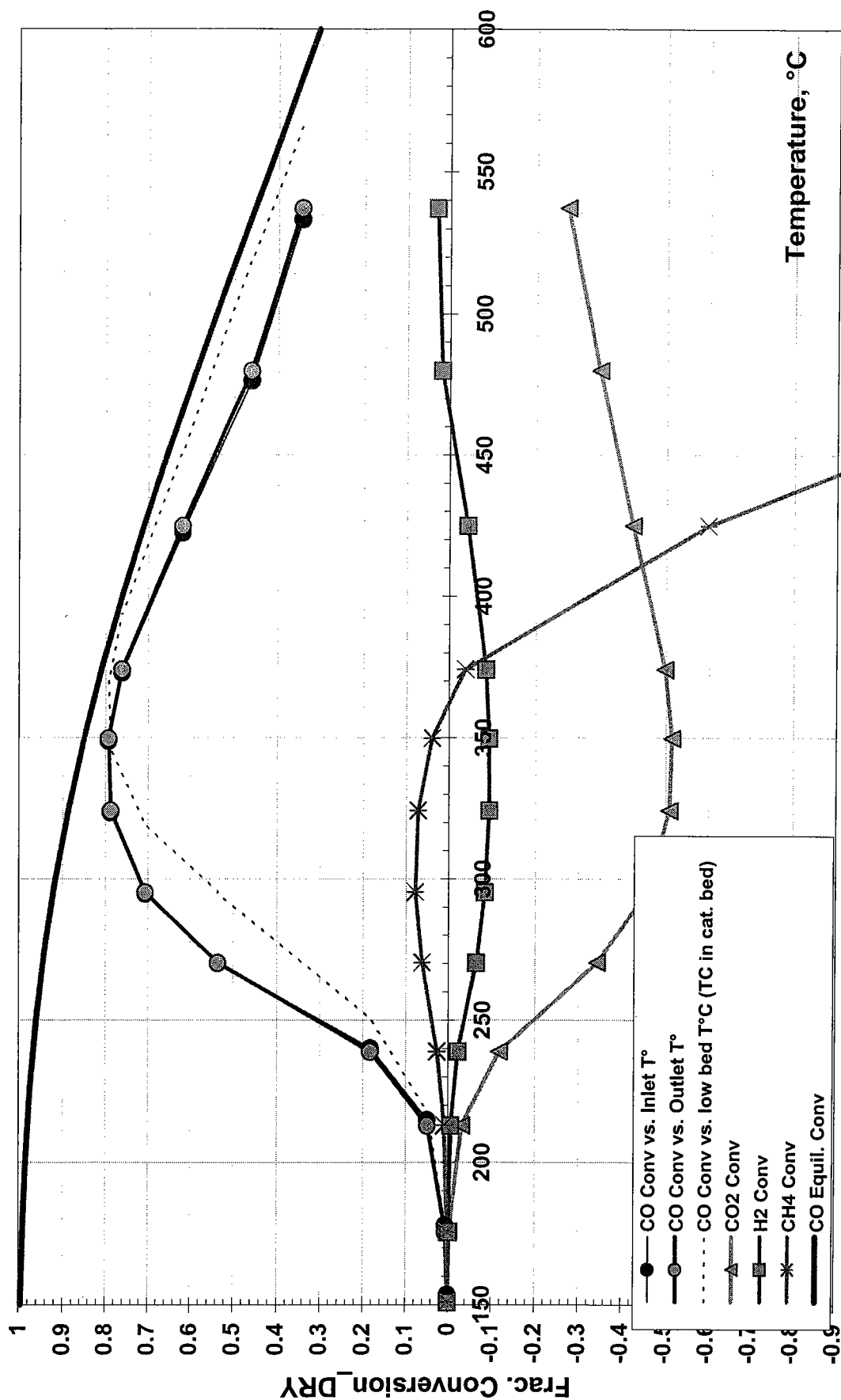




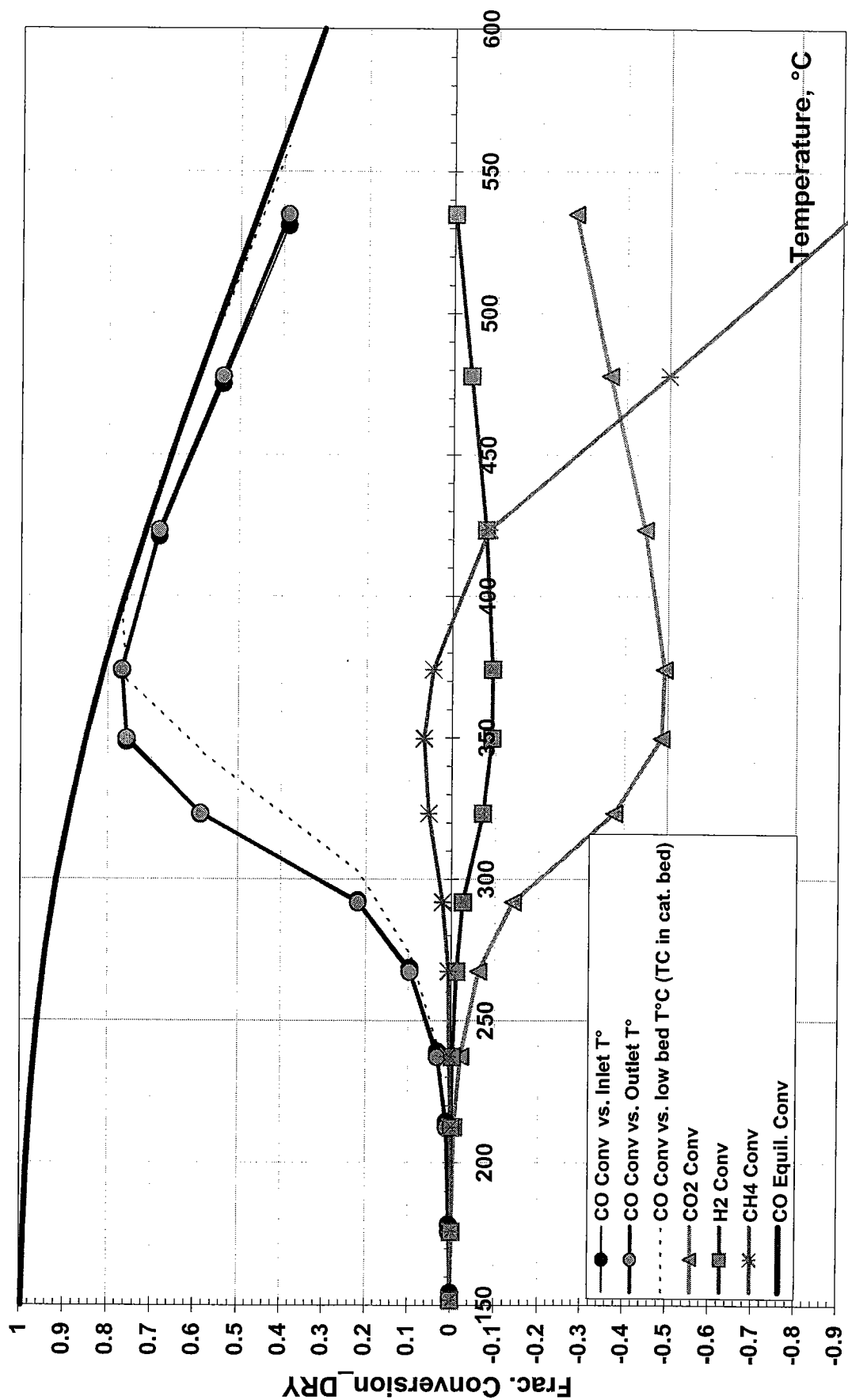
FPR148 0.2%Cs/2%Pt/CeO2-ZrO2 () (1757-163B) 1.0gcat/1.0g cord, 45-60 mesh,
Total Flow 1.125slpm



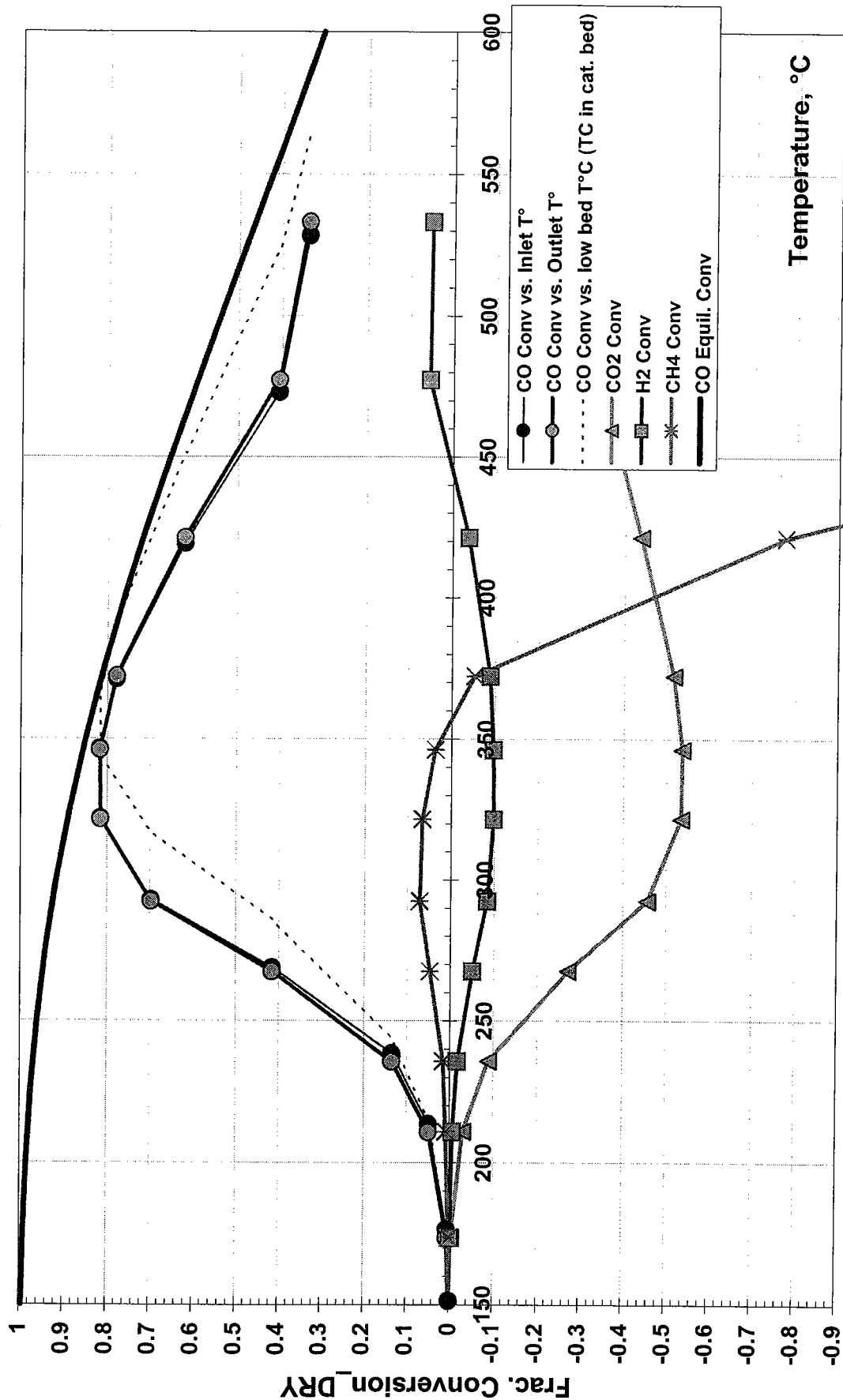
FPR150 0.2%Cs/2%Pt/CeO₂-ZrO₂ () (1757-163B) 1.0gcat/1.0g cord, 45-60 mesh,
Total Flow 1.125slpm



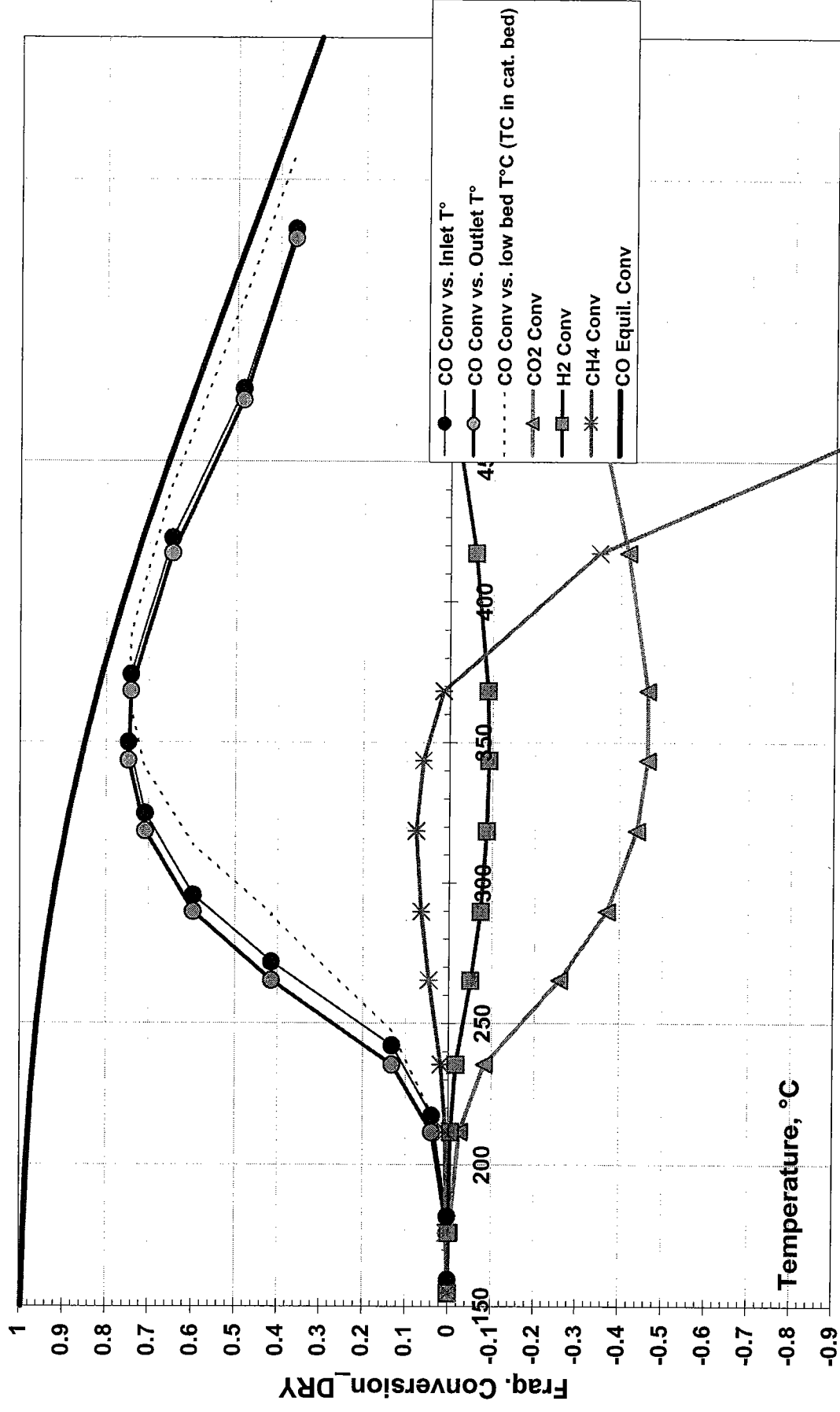
FPR152 0.2%CsHPA/2%Pt/La-Ce2O3 () (1875-10) 1.0gcat/1.0g cord, 45-60 mesh,
Total Flow 1.125slpm

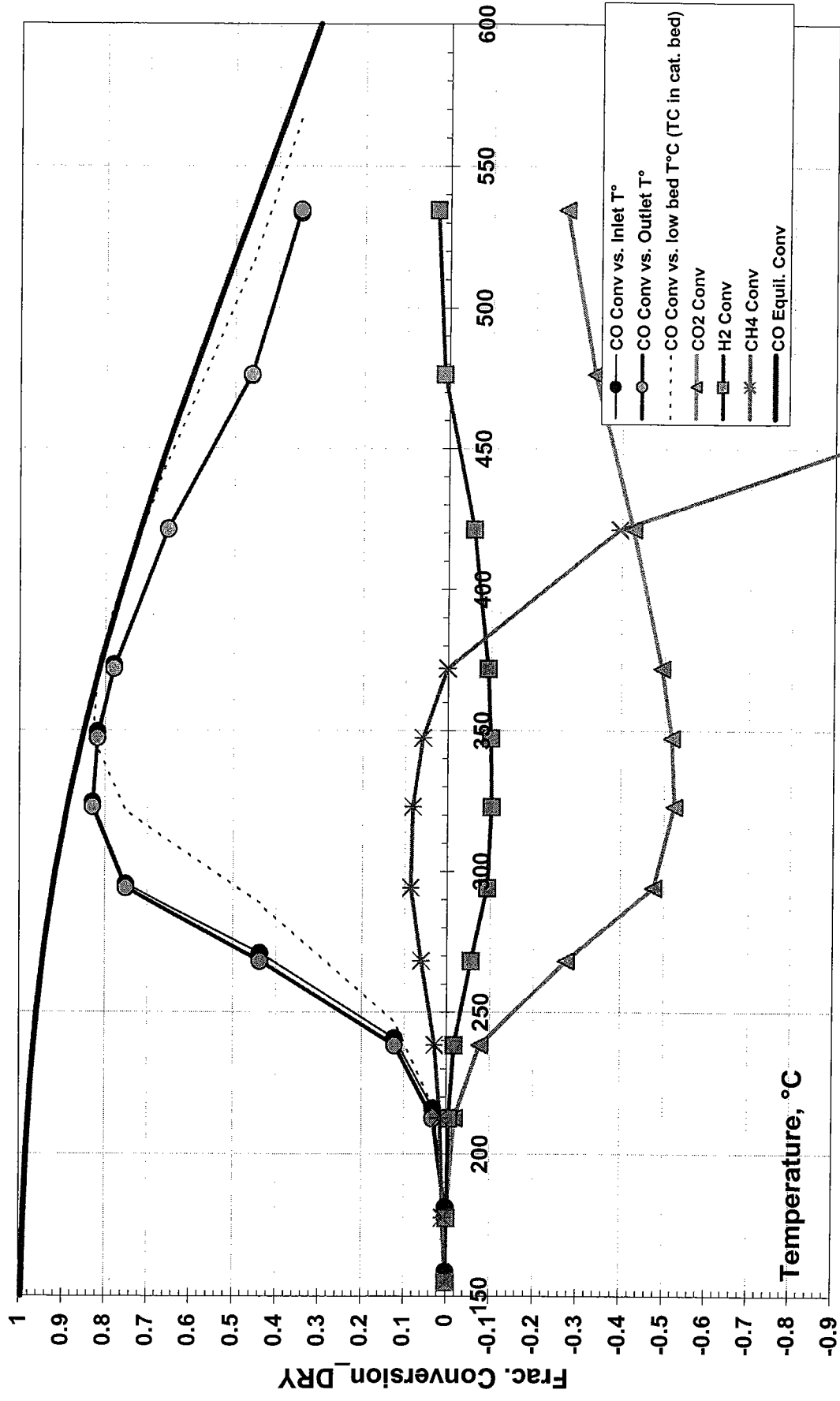


FPR156 2%Pt/CeO₂-ZrO₂ () w/citric acid (1757-166A) 1.0gcat/1.0g cord, 45-60 mesh, Total Flow 1.125slpm



FPR157 (2%Pt-0.2%Cs)/CeO₂-ZrO₂ (), co imp. DI (1757-166B) 1.0gcat/1.0g cord,
45-60 mesh, Total Flow 1.125slpm





Catalyst Compositions

La₂O₃/CeO₂

<u>Catalyst ID</u>	<u>Test Number</u>	<u>Catalyst Composition</u>
1757-131-1	FPR 96	1%Pd/75%La ₂ O ₃ -25%CeO ₂
1757-131-2	FPR 97	1%Pd-0.5%Pt/75%La ₂ O ₃ -25%CeO ₂
1757-133-4	FPR 99	1%Pt-0.5%Cs/75%La ₂ O ₃ -25%CeO ₂
1757-139-2	FPR 104	1%Pt/75%La ₂ O ₃ -25%CeO ₂
1757-136-1	FPR 105	1%Pd/25%La ₂ O ₃ -75%CeO ₂
1757-137-2		1%Pd/0.5%Pt/25%La ₂ O ₃ -75%CeO ₂
1757-141-1	FPR 107	1%Pt-0.5%Cs/ 25%La ₂ O ₃ -75%CeO ₂
1757-140-1	FPR 108	1%Pt/ 25%La ₂ O ₃ -75%CeO ₂
C480-96A	FPR 123	1%Pt/ %La ₂ O ₃ -%CeO ₂
1757-159B	FPR 141	2%Pt/ 9%La ₂ O ₃ -91%CeO ₂
1757-160B	FPR 143	2%Pt/0.2%Cs/9%La ₂ O ₃ -91%CeO ₂
1757-162-A	FPR 144	0.5%Rh/9%La ₂ O ₃ -91%CeO ₂
	FPR 152	0.2%CsHPA/2%Pt/9%La ₂ O ₃ -91%CeO ₂

CeO₂/ZrO₂

<u>Catalyst ID</u>	<u>Test Number</u>	<u>Catalyst Composition</u>
C480-112A	FPR 135	Li-0.5%Rh/58%CeO ₂ -42%ZrO ₂
C480-112B	FPR 139	Cs,0.5%Rh/58%CeO ₂ -42%ZrO ₂
C480-112C	FPR 138	Cs,1%Rh/58%CeO ₂ -42%ZrO ₂
1757-161B	FPR 146	2%Pt/58%CeO ₂ -42%ZrO ₂
1757-162B	FPR 148	2%Pt/0.2%Cs/58%CeO ₂ -42%ZrO ₂
1757-163B	FPR 150	2%Pt/0.2%Cs/58%CeO ₂ -42%ZrO ₂
1757-166A	FPR 156	2%Pt/58%CeO ₂ -42%ZrO ₂
1757-166B	FPR 157	2%Pt/0.2%Cs/58%CeO ₂ -42%ZrO ₂
1757-167B	FPR 159	2%Pt/0.2%Cs/58%CeO ₂ -42%ZrO ₂